# Understanding Your Water Source For Use In Recycled Aquaculture Systems

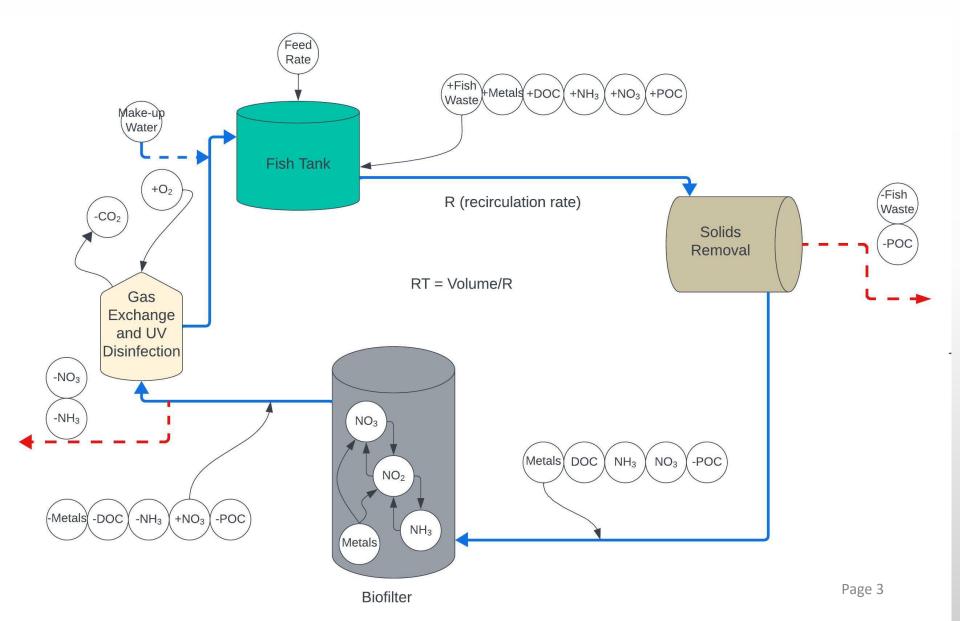
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#### Water Source: Important Questions

- Is the source suitable for raising fish?
  - 1. Correct Quality or Correctable Quality: pH, temperature, clarity, mineral content, etc.
  - 2. Enough Quantity
  - 3. This is a fish biologist concern specific to species
- Is the source suitable for RAS?
  - 1. Will water source support ammonia removal through controlled nitrification in a bioreactor?
  - 2. This question is rarely asked nitrification is generally assumed to occur with a "hot-start"
  - 3. Source water quality can become a limiting issue for maximum yield

### **Basic RAS Diagram**



### Primary RAS Need: Nitrification

- Nitrification = 3 step biofilm process to convert ammonia (fish waste) to nitrate in a bioreactor
- Step 1: Conversion of Ammonium to Hydroxyl Amine (HA)
  - 1. Performed by Ammonia Oxidizing Bacteria (AOB)
  - 2. Requires oxygen and copper
  - 3. Requires acidity so that ammonia is ammonium
  - 4. Creates acidity so that HA is neutral charge
  - 5. HA is neutral charge at pH < 4 su
  - 6. AOB are negatively charged
  - 7. Biofilm needed to trap acidity to keep HA as an acid and make available to AOBs

### Primary RAS Need: Nitrification

- Step 2: Conversion of HA to nitrous acid (HNO<sub>2</sub>)
  - 1. Performed by Ammonia Oxidizing Bacteria (AOB)
  - 2. Requires alkalinity, phosphorus, iron and zinc
  - 3. Creates additional acidity
  - 4. HNO<sub>2</sub>
    - a. neutral at pH < 1.3 su
    - b. 50% neutral at pH = 3.3 su
  - Stable biofilm is needed to trap acidity to keep HNO<sub>2</sub> as an acid and make available to the next step performed by Nitrite Oxidizing Bacteria (NOB)
  - 6. This step is highly corrosive to metal and why plastics are used in RAS

### Primary RAS Need: Nitrification

- Step 3: NO<sub>2</sub> to HNO<sub>3</sub>
  - 1. Performed by Nitrite Oxidizing Bacteria (NOB)
  - 2. Requires alkalinity, phosphorus, <u>iron</u>, <u>molybdenum</u> and <u>sulfur</u>
  - 3. HNO<sub>3</sub>
    - a. neutral at pH < -3.3 su
    - b. 99% anionic at pH = 0.3 su
  - 4. Biofilm will push this anion out of the biofilm via repulsion

### Water Quality Requirements for RAS

- Source with low nitrate desired self seeding
- Low to moderate pH and low ammonia.
  - 1. At pH<7.25, 99% of ammonia waste is present as ammonium ( $NH_4$ +)
  - 2. AOB are negatively charged and will attract NH4+
  - 3. Ammonia in source water may overload RAS treatment capacity
- Low dissolved organic carbon (DOC)
  - 1. DOC creates competition for ammonia
  - 2. DOC disrupts biofilm to cause release of acidity, HA and HNO<sub>2</sub> to disrupt nitrification
- Adequate metals and sulfur prevent inhibition of nitrification process

### **Basic RAS Operation**

- Hydraulics
  - 1. Volume of system, V (gallons or Liters)
  - 2. Recirculation rate of system, R (gal/min, L/min)
  - 3. Recycle Time, RT = V/R (minutes)
- Loading
  - 1. Feed Rate, FR (Kg/day)
  - 2. Ammonia produced per day, APR (grams/day) APR = FR \* %waste
  - 3. Ammonia at steady state, AS (mg/L) AS = APR\*1000/(R\*60\*24)
  - 4. Back calculate % waste from AS

## Troubleshooting RAS

- Measure pH, ORP, conductivity, nitrite, nitrate and ammonia. Microscopy of media helpful.
- Nitrite curve followed by nitrate curve at startup
  - 1. Seeding drip feed ammonia at a steady rate
  - 2. When the rate of ammonia increase slows down, nitrification is starting.
  - 3. As nitrite drops, nitrate should increase.
  - 4. May need to drip metals to assist seeding/startup.
- Climbing and then steady nitrate = desired
- Climbing ammonia and nitrite = undesired
  - 1. Caused by unsteady state, biofilm disruption, change is source quality, competition, or Inhibition
  - 2. MBBR may be too small for feed rate

## Troubleshooting RAS

- Declining pH with steady ammonia and nitrite
  - 1. Sulfur oxidation
  - 2. Minor biofilm instability
- No ammonia or nitrite
  - 1. super stable biofilm
  - 2. Reduced feeding
- Poor water clarity
  - 1. HRT too long or feed rate to high solids breakdown
  - 2. Slime production
  - 3. If amoeba present = recycle time too long
- Source water changes biofilm in well
  - 1. Clarity
  - 2. Odor



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#### Simple Source Assessment

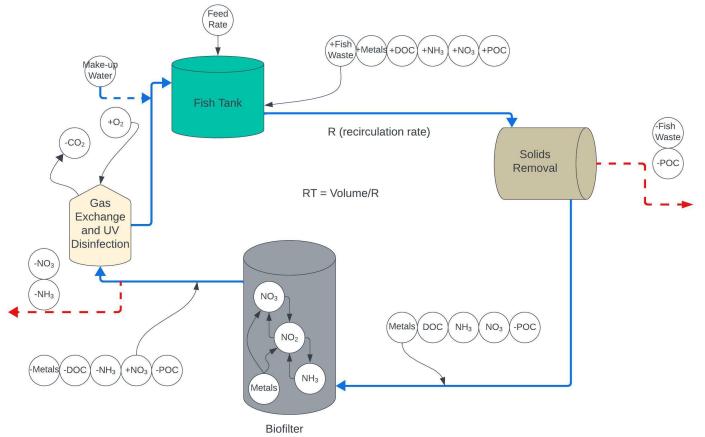


As Collected

After Bleach Addition

Well appears to have corrosive biofilm

### Questions?



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