Student Syllabus for Chemistry 106-02 Fundamental Chemistry Fall 2022

Instructor: Dr. David Szpunar

Office: Chemistry Biology Building 406

e-mail: dszpunar@uwsp.edu
Phone: 715-346-2888

Lecture: Mon., Tue., Thurs. 3:00 PM-3:50 PM Location: CBB 105

Discussion: 02D1 Tues. 1:00 PM-1:50 PM **Location:** CBB 105

Lab: 02L1 Tues. 8:00AM-10:50AM Location: CBB 236 (Dr. Lawrence)

02L2 Thurs. 8:00AM-10:50AM CBB 236

02L3 Fri. 11:00AM-1:50PM CBB 236 (Dr. Cole)

02L4 Wed. 2:00PM-4:50PM CBB 236

Text: <u>Lecture</u>: <u>Chemistry: Structure and Properties, 2nd ed</u>., Nivaldo J. Tro; Pearson: Hoboken,

New Jersey, 2018 (required)

• *ISBN-13*: 978-0134293936

• *ISBN-10*: 0134293932

• This is available for rental at the University Bookstore

<u>Achieve Online Homework</u> (required)

- Can login/create an account at: https://achieve.macmillanlearning.com/start
- MacMillan Achieve trouble-shooting help can be found here: <u>Home (force.com)</u> or call 1.800.936.6899

Scientific calculator (required)

<u>Lab</u>: <u>Labflow Laboratory Package</u> (required)

- Available for purchase at the University Bookstore
- OR visit https://labflow.com/app/login

<u>Laboratory Notebook</u> (required)

 A simple bound composition notebook available for purchase at the University Bookstore

Laboratory safety goggles (required)

• Available for purchase at the University Bookstore

Office hours: <u>Dr. Szpunar</u>

Mon. 11:00AM-12:00 PM Wed. 11:00AM-12:00 PM Fri. 9:00AM-10:00AM

By appointment

COURSE DESCRIPTIONS (CHEM 105 & CHEM 106)

CHEM 105 - Fundamental Chemistry (5 cr)

Description:

(Two semester basic course) Fundamental principles and theories of chemistry, including stoichiometry, atomic and molecular structure and bonding, nuclear chemistry, thermodynamics, descriptive chemistry of nonmetals and transition metals, chemical kinetics and equilibria, introduction to organic chemistry.

Prerequisites:

MATH 090 or placement in MATH 100 or above. (See notes 3, 4, 5 in Course Catalog).

CHEM 106 - Fundamental Chemistry (5 cr)

Description:

Continuation of CHEM 105.

Prerequisites:

CHEM 105, MATH 100 or higher. (See note 4 in Course Catalog.)

Note that MATH 100 is being phased-out and two courses, MATH 95 and MATH 107, have been introduced to replace it.

COURSE OBJECTIVES

General Education Program (GEP)

Natural Sciences (Learning Outcomes)

- To be able to explain major concepts, methods, or theories in the natural sciences to investigate the physical world.
- To be able to interpret information, solve problems, and make decisions by applying natural science concepts, methods, and quantitative techniques.
- (lab component also required)
- To be able to describe the relevance of aspects of the natural sciences to their lives and society.

Student Learning Outcomes (Chemistry Department)

Students graduating with a major in Chemistry from the University of Wisconsin-Stevens Point will be able to perform tasks representing all eight of the following learning outcomes. Students completing Chem 106 will perform tasks, at an introductory level, representing the <u>underlined</u> learning outcomes.

- Apply the foundational principles of the major sub disciplines of chemistry (Analytical, Inorganic, Organic, and Physical) to identify and explain the chemical and physical properties of matter.
- Demonstrate safe work habits and techniques in the chemical laboratory
- Analyze experimental results in order to draw justifiable conclusions
- Evaluate, document, and communicate experimental results and chemistry related issues according to accepted scientific standards, as a written report, as a poster, and as an oral presentation
- Design and construct experiments to address chemical problems using appropriate methods, techniques, equipment, and modern instruments for the synthesis, isolation, and characterization of matter and the analysis of mixtures using relevant information compiled from scientific literature.

Living and Learning During COVID-19

University COVID-19 Information Page:

• The University COVID-19 Information page can be found here: Home - Information on Coronavirus | UWSP

Face Coverings:

- At all UW-Stevens Point campus locations, the wearing of face coverings is mandatory in all buildings, including classrooms, laboratories, studios, and other instructional spaces. Any student with a condition that impacts their use of a face covering should contact the <u>Disability and</u> <u>Assistive Technology Center</u> to discuss accommodations in classes. Please note that unless everyone is wearing a face covering, in-person classes cannot take place. This is university policy and not up to the discretion of individual instructors. Failure to adhere to this requirement could result in formal withdrawal from the course.
- A document featuring the UWSP face covering guidelines is posted on our class Canvas site.

Other Guidance:

- Please monitor your own health each day using this screening tool. If you are not feeling well or believe you have been exposed to COVID-19, do not come to class; email your instructor and contact Student Health Service (715-346-4646).
 - As with any type of absence, students are expected to communicate their need to be absent and complete the course requirements as outlined in the syllabus.
- Maintain a minimum of 6 feet of physical distance from others whenever possible.
- Do not congregate in groups before or after class; stagger your arrival and departure from the classroom, lab, or meeting room.
- Wash your hands or use appropriate hand sanitizer regularly and avoid touching your face.
- Please maintain these same healthy practices outside the classroom.

METHODOLOGY: The class is composed of three lectures, one lab, and one discussion meeting per week. The lecture itself will primarily be lecture-based, with three 50-minute lecture exams (total of 375 points/37.5% of the grade). The discussion section is usually earmarked for Q&A sessions and group work. Unless otherwise noted, you will perform laboratories weekly. Please see the laboratory schedule provided in this syllabus.

Canvas: Course information (quiz solution sets, course grades, lecture slides etc.) will be posted daily on Canvas. It is your responsibility to visit the site daily. You can log into Canvas at: https://www.uwsp.edu/canvas/Pages/default.aspx

Exams: There will be three hourly exams (total of 375 points/37.5% of the grade) will be given on Thurs., Oct. 6; Tues., Nov. 1; and Thurs. Dec. 1.

Quizzes: There will be eight 10-15 minute quizzes at the start of discussion (see schedule for dates) each worth 10 points for a total of 80 points/8% of the grade.

Final Exam: Note that the final exam is cumulative covering Ch. 11, 13-19, and is worth a total of 200 points (29% of the grade). It is scheduled for Monday, 12/19 8:00-10:00AM in CBB 105

Discussion Group Assignments: These problems will be turned in at the end of discussion and are worth a total of $8 \times 4 = 32$ points (3.2% of your grade).

Homework: We will be using the Achieve online homework system. Mandatory homework is assigned for each chapter. There are a total of eight assignments (20 pts each), giving a total of 160 points/16% of grade. The due dates are given in tabular format at the end of the syllabus. It is essential to do the homework in order to succeed in this class. Exams and quizzes are heavily based on homework. You may work together on the homework, but each student is responsible for understanding each problem. Copying another's homework is not "group work" - it is plagiarism. If asked, you are responsible for justifying that your work is your own. I will not credit anything that I perceive to be copied work.

Laboratory: As detailed in the laboratory schedule, eleven experiments will be assigned (including the safety quiz). Labs are worth 17 points each. I will drop your two lowest lab scores, giving a total of 153 points/15.3% of grade. Prelabs (called "prelab quizzes" in LabFlow) are due by the Sunday before the lab is to be performed by 11:59 PM. Prelabs will be worth 3 points of the total 17 points for each lab. You are also required to prepare your lab notebook for each upcoming lab. An example of the lab notebook grading scheme is shown on the next page. All reports and copies of lab notebook pages are submitted using LabFlow and are due by the following Monday at 11:59 PM. (Please see Canvas for due dates.) **There are no makeup labs**.

Attendance: Attendance for all lectures, discussions and laboratories is expected as outlined in the UWSP Undergraduate Catalog. See the section about Attendance under Academic Policies. More than 3 unexcused absences will result in a loss of 3 attendance points (.3%).

<u>Laboratory notebook grading rubric</u>:

CHEM 105 Lab Notebook Grading Rubric Name:	
Experiment: Section	tion:
Item	$\sqrt{}$
To be Completed by the Lab Instructor (pre-lab):	
Updated table of contents	
Experiment title	
Experiment purpose	
Brief procedure or flow chart	Score on post lab
Data tables prepared in advance	questions
All entries made in ink	Lab notebook
Lab Instructor Signature/initials:	deductions
To be Completed by the Lab Grader (post-lab):	Overall Lab Score
Date and signature present at the bottom of each page with data	a
All data present in tables with titles, headings, and units	
Data errors appropriately labeled and corrected	
Results summary and/or conclusion	
Total Number of missing/incorrect Items	

Notes: $\sqrt{\text{means item is present and correct.}}$

0.25 pts will be deducted for each missing/incorrect item

General considerations:

- -Significant figures will be considered on all quizzes, exams, lab reports, prelabs and homework assignments.
- -Phases (i.e. solid, liquid, gas or aqueous) must be included in all chemical equations.
- -Carbon atoms are shown with more than 4 bonds in Lewis structures etc.

Make-up policy: There will be **NO** unexcused make-ups of homework, exams, or quizzes. Any excused makeups must be performed within 48 hours of the original date. Due to the nature of labs, there are no lab makeups. However, your two lowest 17-point labs will be dropped to compensate for this. All unexcused absences will result in a grade of zero.

Grading: Your final grade will be based on the following point system:

Hourly exams:	$3 \times 125 \text{ pts}$	=	375 points	(37.5%)
Final exam:	$1 \times 200 \text{ pts}$	=	200 points	(20%)
Discussion Quizzes:	$8 \times 10 \text{ pts}$	=	80 points	(8%)
Syllabus Quiz:	$1 \times 10 \text{ pts}$	=	10 points	(1%)
Homework:	$8 \times 20 \text{ pts}$	=	160 points	(16%)
Lab:	10 pts + 10×13 pts	=	140 points	(14%)
Discussion group problem sets:	8×4 pts	=	32 points	(3.2%)
Attendance:	3 pts	=	3 points	(.3 %)
•				

1000 points (100%)

You will be graded on the following scale:

% Total Points	Grade	% Total Points	Grade
≥93 %	A	73 – 76 %	C
90 – 92 %	A-	70 – 72 %	C-
87 – 89 %	B+	67 – 69 %	D+
83 – 86 %	В	63 – 66 %	D
80 – 82 %	B-	< 63 %	F
77 – 79 %	C+		

Lecture policies:

You are expected to be at class on time. There will be no make-up quizzes.

Attendance for all lectures, discussions and laboratories is expected as outlined in the UWSP Undergraduate Catalog. See the section about Attendance under Academic Policies.

UWSP is committed to providing reasonable and appropriate accommodations to students with disabilities and temporary impairments. If you have a disability or acquire a condition during the semester where you need assistance, please contact the Disability and Assistive Technology Center on the 6th floor of Albertson Hall (library) as soon as possible. DATC can be reached at 715-346-3365 or DATC@uwsp.edu.

Bring your text, a calculator, and note-taking materials to every class. This is important because we will often times need calculators and books for group work. I will not supply "loaner" calculators--you must bring your own.

Please do not hesitate to raise your hand and ask questions during lecture if you are unclear on some point.

You are responsible for checking your e-mail and Canvas daily.

Working in groups is encouraged. However, **copying work for homework assignments is** unacceptable. Any such assignments will not be accepted and will receive a score of zero points. This includes lab results as well!

Please turn all cell phones to vibrate before class. No texting or iPods allowed. Laptops are for taking notes only. If I see you texting or using your laptops in an inappropriate manner I will give you one warning before asking you to leave the class. *Talking/texting/social media is inappropriate and will not be tolerated. It distracts other students and is rude.*

Treat all fellow students with respect and civility. Failure to do so will result in your dismissal from that day's lecture.

Study Tips for General Chemistry.

- Lectures will primarily follow the text. Read the assigned sections before lecture and again shortly after, using your lecture notes as a supplement. Repeating the information helps the facts and concepts sink and remain in your brain.
- Try to work the problems as soon as possible after lecture. This will help you discern where your understanding may be lacking and will help reiterate the important concepts.
- I cannot overemphasize the importance of peer groups! Form a group of students from class and plan to meet outside of class at least once a week to discuss problems and material. Your peers may have picked up on something in lecture that you missed, they may be able to explain something in terms that you will understand better than I may be able to do, and you will be reviewing the material which will again help you to remember it come test time.
- This course covers some difficult material and necessarily maintains a rapid pace. Expect to spend 9-12 hours per week outside class for study, homework and writing lab reports. If you cannot commit to this level of study, I recommend you reduce your other commitments or withdraw from the course.
- Do the homework, re-do the homework, and do practice problems. "Practice makes perfect."

Academic Honesty/Plagiarism Policy:

You are encouraged to study together, work problems and exercises with others in the class, and to seek help in understanding the material. However, unless specifically instructed otherwise, all work to be graded should be your own work, and not copied from any other person. Any instances of plagiarism or cheating will be dealt with in accordance with the UWSP Chapter 14 rules on Academic Misconduct. Any violations will result in a zero for that assignment/exam. A second violation results in an F for a final grade in the class.

Accommodation of Persons with Disabilities:

The Americans with Disabilities Act (ADA) is a federal law requiring educational institutions to provide reasonable accommodations for students with disabilities. If you have a disability and require classroom or exam accommodation, please register with the Disabilities Services office and then contact me within the first two weeks of the semester. In order to receive accommodations, you must have documentation of your disability on file with the Office of Disability Services. In addition, you must provide me with an Accommodations Request Form (available on their website). You must have me sign the form and return it to the Office of Disability Services.

University Policy on Absence to Observe Religious Holidays

It is UW System policy to reasonably accommodate your sincerely held religious beliefs with respect to all exams and other academic requirements. You will be permitted to make up an exam or other academic requirement at another time or by an alternative method, without any prejudicial effect, if:

- There is a scheduling conflict between your sincerely held religious beliefs and taking the exam or meeting the academic requirements; and
- You have notified me within the first three weeks of the beginning of classes of the specific days or dates that you will request relief from an examination or academic requirement.
- I will accept the sincerity of your religious beliefs at face value and keep your request confidential.
- I will schedule a make-up exam or requirement before or after the regularly scheduled exam or requirement.
- You may file any complaints regarding compliance with this policy in the Equity and Affirmative Action Office.

Tentative Lecture Outline (Subject to Change)

	(Subject to Change)			
Week	Day	Date	Lecture	
	M	5-Sep	Labor Day—No Class	
1	T	6-Sep	Introduction/Ch. 11	
	Th	8-Sep	Ch. 11	
	M	12-Sep	Ch. 11	
2	Т	13-Sep	Ch. 11	
	Th	15-Sep	Ch. 11	
	M	19-Sep	Ch. 11	
3	T	20-Sep	Ch. 11	
	Th	22-Sep	Ch. 13	
	M	26-Sep	Ch. 13	
4	T	27-Sep	Ch. 13	
	Th	29-Sep	Ch. 13	
	M	3-0ct	Ch. 13	
5	T	4-0ct	Ch. 14	
	Th	6-0ct	Exam #1 (Ch. 11,13)	
	M	10-0ct	Ch. 14	
6	T	11-0ct	Ch. 14	
	Th	13-0ct	Ch. 14	
	M	17-0ct	Ch. 14	
7	T	18-0ct	Ch. 15	
	Th	20-0ct	Ch. 15	
	M	24-0ct	Ch. 15	
8	T	25-0ct	Ch. 15	
	Th	27-0ct	Ch. 15	
	M	31-0ct	Ch. 16	
9	Т	1-Nov	Exam #2 (Ch. 14-15)	
	Th		Ch. 16	
	M	7-Nov	Ch. 16	
10	Т	8-Nov	Ch. 16	
	Th	10-Nov	Ch. 16	
	M	14-Nov	Ch. 16	
11	T	15-Nov	Ch. 17	
	Th	17-Nov	Ch. 17	
	M	21-Nov	Ch. 17	
12	T	22-Nov	Ch. 17	
	Th	24-Nov	Thanksgiving: No Class	
	M	28-Nov	Ch. 17	
13	T	29-Nov	Ch. 18	
	Th	1-Dec	Exam #3 (Ch. 16-17)	
	M	5-Dec	Ch. 18	
14	T	6-Dec	Ch. 18	
	Th	8-Dec	Ch. 18	
	M	12-Dec	Ch. 19	
15	T	13-Dec	Ch. 19	
	Th	15-Dec	Ch. 19	
	M	19-Dec	FINAL EXAM	
×	T	20-Dec		
	Th	22-Dec	15 1 1 1 1	

Please note, the last day to drop without a grade is Thurs, Sept. 15, and the last day to drop a course is Fri., Nov. 11.

Tentative Discussion Outline (Subject to Change)

Week	Day	Date	Discussion
1	Tue	6-Sep	Introductions
2	Tue	13-Sep	Ch. 11 Group Problems
3	Tue	20-Sep	Quiz #1 (Ch. 11)/Ch. 11 Group Problems
4	Tue	27-Sep	Quiz #2 (Ch. 13)/Ch. 13 Group Problems
5	Tue	4-0ct	Group Problems/Exam preparation
6	Tue	11-0ct	Quiz #3 (Ch. 14)/Ch. 14 Group Problems
7	Tue	18-0ct	Ch. 15 Group Problems
8	Tue	25-0ct	Quiz #4 (Ch. 15)/Ch. 15 Group Problems
9	Tue	1-Nov	Ch. 16 Group Problems
10	Tue	8-Nov	Quiz #5 (Ch. 16)/Ch. 16 Group Problems
11	Tue	15-Nov	Ch. 17 Group Problems
12	Tue	22-Nov	Quiz #6 (Ch. 17)/Ch. 17 Group Problems
13	Tue	29-Nov	Thanksgiving: Campus Closed
14	Tue	6-Dec	Quiz #7 (Ch. 18)/Ch. 18 Group Problems
15	Tue	13-Dec	Quiz #8 (Ch. 19)/Ch. 19 Group Problems

Tentative Homework Due Dates (Subject to Change)

(Bub)eet to dhange)				
Day	Date	Due		
Thurs.	22-Sept	HW #1 (Ch. 11)		
Tues.	4-0ct	HW #2 (Ch. 13)		
Tues.	18-0ct	HW #3 (Ch. 14)		
Mon.	31-0ct	HW #4 (Ch. 15)		
Tue.	15-Nov	HW #5 (Ch. 16)		
Tue.	29-Nov	HW #6 (Ch. 17)		
Mon.	12-Dec	HW #7 (Ch. 18)		
Mon.	19-Dec	HW #8 (Ch. 19)		

All homework assignments due at 11:59PM on the indicated date

Lab Schedule*

		- •
Week	Dates	Lab
1	9/5-9/9	Check-in/Online Safety Lab
2	9/12-9/16	Lab #1: Modeling, Geometry, and Polarity
3	9/19-9/23	Lab #2: Intermolecular forces
4	9/26-9/30	Lab #3: Solutions, Electrolytes, and Concentrations
5	10/3-10/7	Lab #4: Molar Mass from Freezing Point Depression
6	10/10-10/14	Lab #5: Glassware, Techniques, and Measurement
7	10/17-10/21	Lab #6: Iodine Clock Reaction
8	10/24-10/28	Lab #7: Le Châtelier's Principle
9	10/31-11/4	Lab #8: Determination of K _{sp}
10	11/7-11/11	Lab #9: Amount of NaOCl in Bleach
11	11/14-11/18	Lab #10: Titration of a Diprotic Acid
12	11/21-11/25	No lab: Thanksgiving
13	11/28-12/2	Lab #11: Buffers
14	12/5-12/9	Lab #12: Voltaic Cells
15	12/12-12/16	Check-out

*Prelabs are due by 11:59 PM on the Sunday before the lab is to be performed. Notebook pages and lab reports are due by 11:59 PM on the Sunday after the lab is to be performed.